NOSB NATIONAL LIST FILE CHECKLIST

PROCESSING

MATERIAL NAME: Potassium chloride			
CATEGORY: Synthet	tic Allowed Complete?: 3/17		
	NOSB Database Form		
	References		
	MSDS (or equivalent)		
	FASP (FDA)		
	Date file mailed out:1/8/95		
	TAP Reviews from: Bob Ours + Richard 74-ver		
	Richard Thourr		
	Supplemental Information:		

MISSING INFORMATION:	

NOSB/NATIONAL LIST COMMENT FORM/BALLOT

Use this page to write down comments and questions regarding the data presented in the file of this National List material. Also record your planned opinion/vote to save time at the meeting on the National List.

Name of Material	Potassi	um chlori	<u>de</u>
Type of Use:	Crops;	Livestock;	✓ Processing
TAP Review by:			
1. Steve	Taylor		
2. Riche	ed Theor	<u>r</u>	
3. <u>B.b</u>	Ourst		
Comments/Question	is:		
My Opinion/Vote i	S:		
Signature		Date	•

Non

allower

USDA/TAP REVIEWER COMMENT FORM

Use this page or an equivalent to write down comments and summarize your evaluation regarding the data presented in the file of this potential National List material. Attach additional sheets if you wish.

This file is	s due back to us wit	thin 30 days of:	Jan 7.
Name of I	Material: Potassi	um Chloride	
Reviewer	Name: Siwe	Taylor	
is this sub	Synthetic (may be possible to mai	
Please comm	nent on the accuracy of	the information in the	o file:
This mater	ial should be added	to the National Lie	et ac
	Synthetic Allowed	•	d Natural
or, List becaus	This material doese:	s not belong on ti	ne National
Are there of the control of the cont	any restrictions or li this material by use ist?	mitations that shou or application on t	ild be the
Any additio	onal comments or ref	ierences?	
Signature	Shore Taylor	Date 3-5°	-9

3			

USDA/TAP REVIEWER COMMENT FORM

Original mailing date: 7 Jan 1995.

Name of Material: Potassium Chloride Reviewer Name: Richard C. Theuer

NATURAL Potassium chloride (muriate of potash) is produced from brines in the United States.

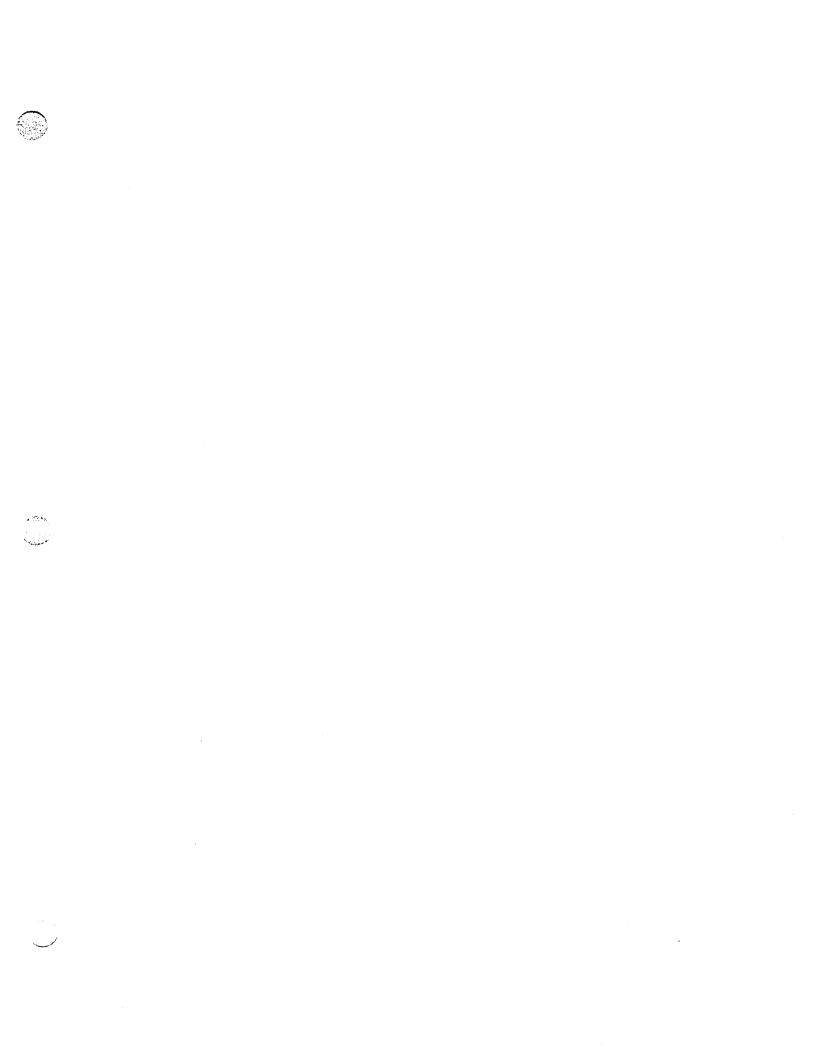
COMMENTS RE SECTION 2119 (m) CRITERIA:

- 1. Potassium and chloride are nutritionally essential elements for man and other animals.
- 2. Potassium salts have a bitter taste so use is self-limiting. Potassium chloride is used as a salt substitute, especially for salt-sensitive hypertensive individuals.
- 3. Alternatives to potassium chloride as a nutrient, dietary supplement or yeast food exist, but virtually all other potassium salts are synthetic.
- 4. The fact that potassium chloride is produced from brine minimizes the environmental impact of its manufacture and makes it more compatible with long-term sustainability.

The following natural substance should be allowed as an ingredient in organic foods. It should not be added to the National List of natural substances prohibited for use as ingredients or processing aids in Organic Food:

potassium chloride.

18 Feb 1995



USDA/TAP Reviewer Comment Form

į	Material: Potassium chloride
	Reviewer: Bob Durst
	Is this substance Natural or Synthetic? Explain (if appropriate) It is a synthetic substance. It is used as a yeast nutrient. It is also a salt substitute, where it is used to lower the sodium content.
	Please comment on the accuracy of the information in the file: The file is accurate.
	This material should be added to the National List as:
	X Synthetic Allowed,
	Prohibited Natural, or
	This material does not belong on the National List because: This substance might be considered for exclution from the list, as it has little (if any) effect for which there is not a suitable substitute (sea salt or sodium chloride).
	Are there any restriction or limitations that should be placed on this material by use or application on the National List?
	Must be listed on the ingredient label.
	As mentioned above, while there are serious health concerns about excess sodium intake in the

As mentioned above, while there are serious health concerns about excess sodium intake in the diet, and the use of KCl as a salt substitute is to reduce this, there are also numerous non-chemical ways to accomplish the same thing. Changes in diet or formulation of products with just reduced salt content (without KCl substitution).

An option might be to limit its use to only those applications where a high salt concentration is necessary to produce the product (ie. pickles, sauerkraut) where it can really show its beneficial effect of reducing sodium content.

Any additional comments or references?

As with all synthetic inorganic salts, source must be food grade. In addition each lot should be analyzed for toxic element concentrations (mercury, lead, cadmium, arsenic, thallium and antimony) and a near zero tolerance adopted.

AU: Freeman,-T.-M. (Tom M.); Gregg,-O.-W. (Owen W.) CA: American Association of Cereal Chemists. Meeting (66th: 1981: Denver, Colo.).

TI: Sodium intake-dietary concerns.

NT: "Based on presentations at the 1981 Flavor and Additives Division's symposium, "Sodium intake--dietary concerns," held October 26, 1981, at the 66th Annual Meeting of the American Association of Cereal Chemists in Denver, Colorado"--Verso t.p.

AB: Abstract: The proceedings of a 1981 symposium for food technologists, nutritionists, and health professionals concerned with the dietary intake of sodium (Na) by consumers and patients presents 9 overview articles prepared by technical authorities from academia, government, industry, and medicine. Topics include: a major review of the nutritional significance of Na (covering about half of the text); concerns of Na intake in medicine and health; an industrial perspective of the role of Na in the diet; regulatory aspects of Na and salt in processed foods; the consumer's perspective of dietary Na; national surveys of individual Na intakes; and functional and nutritional concerns regarding Na levels in cookies and crackers. A performance evaluation concerning the utility of NaCl-KCl mixtures in bread also is included. The text includes a number of data tables and illustrations and numerous literature citations.

TI: Nutrition statements adopted by the National Health and Medical Research Council at its ninety-eighth session (October 1984).

SO: J-Food-Nutr. Canberra, Australia: Australian Government Publishing Service. 1985. v. 41 (4) p. 191-194. charts.

AB: Abstract: Specific recommendations set forth by the Australian National Health and Medical Research Council covering several nutritionally related themes are discussed; viz.: the use of tables of acceptable weight-for-height based, on a body mass index of 20-25; nutrition education for school children; the replacement of NaCl by KCl in food processing; the need for further studies of food intolerance in children; and that breast-feeding should be encouraged. The importance of disseminating the Council's recommendations on breast-feeding, infant nutrition, and infant feeding practices is stressed. Other related recommendations include: the use of a warning label on soy-based infant formulas citing possible intolerance; the labelling of adult and child medications and of foods concerning colorants, additives, sugar, and other ingredients that may cause intolerance; and the use of guidelines for conclusively establishing elimination diets in food intolerance cases.(wz).

AU: Bell,-R.R.; Eldrid,-M.M.; Watson,-F.R. TI: The influence of NaCl and KCl on urinary calcium excretion in healthy young women.

SO: Nutr-Res. Elmsford, N.Y.: Pergamon Press. Jan 1992. v. 12 (1) p. 17-26. CN: DNAL QP141.A1N88

AB: The calciuric effects of NaCl and KCl were compared in 11 healthy young women consuming diets containing 600 mg calcium per day. Each subject participated in three treatment periods: (1) control diet containing 44 mmol Na and 64 mmol K, (2) control diet plus 102 mmol NaCl and (3) control diet plus 104 mmol KCl. Urine was collected during the last two days of each treatment period and analyzed for Ca, Na, K and Cl. The NaCl supplement produced calciuria similar to that reported by other researchers. However, the KCl supplement did not cause calciuria. Subjects receiving the KCl excreted 5.2 +/- 0.4 mmol Ca compared to 7.0 +/- 0.4 mmol Ca during NaCl supplementation and 5.9 +/- 0.6 mmol Ca during the control period. This suggests that the current practice of replacing some NaCl in food processing, or in salt substitutes, with KCl will decrease the amount of sodium consumed by the general population without contributing to calcium excretion. XAU: Curtin University of Technology, Perth, Australia.

Signature Rest Date 3/11/95

NOSB Materials Database

Identification

Potassium chloride Common Name

Chemical Name

Other Names

muriate of potash

Code #: CAS N. L. Category

Synthetic Allowed

Code #: Other

MSDS

yes Ono

Family

Chemistry

Composition

KCI

Properties

Colorless, elongated, prismatic, or cubical crystals, or a white, granular powder. Odorless, has a saline

taste, and is stable in air. Solutions are neutral. Insoluble in alcohol.

How Made

Produced from brines in the United States.

Processing

Use/Action

Type of Use

Specific Use(s) Nutrient, dietary supplement; gelling agent; salt substitute; yeast food.

Action

Combinations

Status

OFPA

N. L. Restriction

EPA, FDA, etc

FDA-GRAS

Directions

Safety Guidelines

State Differences

Historical status

Internation! status

Allowed by IFOAM.

NOSB Materials Database

OFPA Criteria

2119(m)1: chemical interactions

Not Applicable

2119(m)2: toxicity & persistence

Not Applicable

2119(m)3: manufacture & disposal consequences

brine manufacture has little environmental impact.

2119(m)4: effect on human health

None. Potassium is and essential element but is hazardous in high doses; has caused metabolic alkalosis in infants when used as salt substitute in infant formula.

2119(m)5: agroecosystem biology

Not Applicable

2119(m)6: alternatives to substance

Sodium chloride; other potassium salts. Reduce overall salt content in foods with other flavorings etc.

2119(m)7: Is it compatible?

References

See attached.

Also see references in attachment to Bob Durst's TAP review.

POTASSIUM CHLORIDE REFERENCES

AU: Reddy,-K.A.; Marth,-E.H.

TI: Reducing the sodium content of foods: A review.

SO: J-Food-Prot. Ames, Iowa: International Association of Milk, Food, and Environmental Sanitarians. Feb

1991. v. 54 (2) p. 138-150. CN: DNAL 44.8-1824

AB: Salt (sodium chloride), a substance essential for life processes, is the second most-used food additive. It is added to foods as a flavoring or flavor enhancing agent, a preservative, or an ingredient responsible for desired functional properties in certain products. Excessive dietary sodium is believed to contribute to hypertension and development of cardiovascular disease which afflicts ca. 60 million Americans. Common salt substitutes, including potassium chloride, certain herbs, spices, organic acids, autolyzed yeast products, and hydrolyzed vegetable protein products alone or in combination, if used properly, can result in products that are consumer-acceptable.

AU: Tuncan,-E.U.; Martin,-S.E.

TI: Combined effects of salts and temperature on the thermal destruction of Staphylococcus aureus MF-31. SO: J-Food-Sci-Off-Publ-Inst-Food-Technol. Chicago, Ill.: The Institute. May/June 1990. v. 55 (3) p. 833-836.

CN: DNAL 389.8-F7322

AB: The combined effects of salts (NaC1 and KC1) and heat treatment temperature on the thermal destruction of Staphylococcus aureus MF-31 was studied by determining the relationships between the decimal reduction time (D value), temperature and each of three parameters (water activity, osmotic pressure and water binding energy) of the heating menstruum. The results demonstrated that the resistance of S. aureus MF-31 to heat destruction increased as the degree of salt-water association increased. Salts and heat treatment temperature had an interactive effect on heat resistance; however, increasing the heat treatment temperature made the difference caused by salt concentrations become insignificant.

AU: Guy,-E.J.

TI: Effect of KCl levels on functional and baking properties of sponge doughs and hedonic ratings of their

SO: Cereal-Foods-World. St. Paul, Minn.: American Association of Cereal Chemists. June 1986. p. 421-425. CN: DNAL 59.8-C333

AU: Keeton,-Jimmy-T.

TI: Effects of potassium chloride on properties of country-style hams.

SO: J-Food-Sci. Chicago, Ill.: Institute of Food Technologists. Jan/Feb 1984. v. 49 (1) p. 146-148. charts.

CN: DNAL 389.8-F7322

AB: Abstract: A study assessed acceptable levels of KCl as a substitute for NaCl in 16 hams, evaluated weight loss, color changes, and appearance of KCl-cured hams, and characterized the sensory properties of ham slices containing different KCl levels. A 1/3 replacement of NaCl by KCl produced hams having only a slight degree of bitterness, which may be useful for significantly lowering their sodium content.

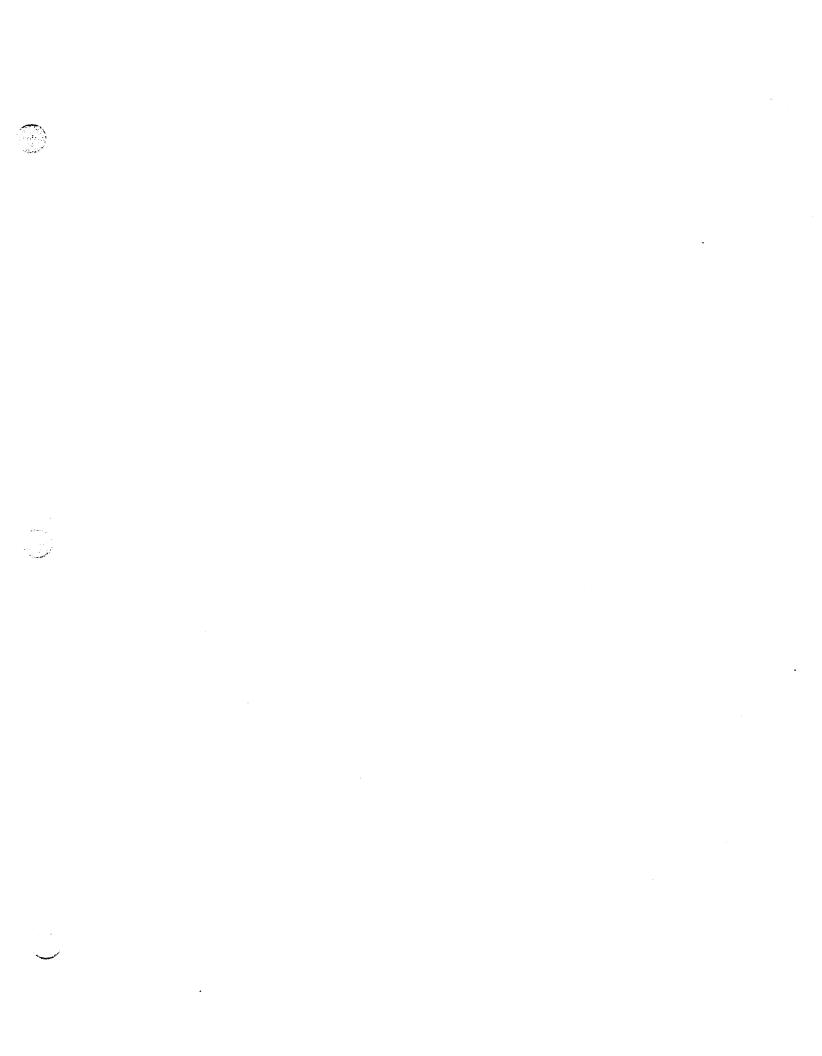
AU: Maurer,-A.J.

TI: Reduced sodium usage in poultry muscle foods.

SO: Food-Technol. Chicago: Institute of Food Technologists. July 1983. v. 37 (7) p. 60-65. charts.

CN: DNAL 389.8-F7398

AB: Abstract: The sodium (Na) content of poultry products, the functions of commercially-added salt in processed meats, and mechanisms for Na reduction in processed poultry products are reviewed and discussed. The research reviewed suggests that sodium chloride (NaCl) is necessary for optimum product flavor, functionality, and safety. Potassium chloride can be substituted for up to 35-50% of the NaCl in foods, but after that amount it adversely affects food flavor.



	DE Page 1
1 - PRODUCT IDENTIFICATION	
PRODUCT NAME: POTASSIUM CHLORIDE FORMULA: KCL CAS NO.: 7447-40-7 COMMON SYNONYMS: POTASSIUM MURIATE PRODUCT CODES: 4920,3047,3052,3046,3040 EFFECTIVE: 05/05/86	FORMULA WT: 74.56 NIOSH/RTECS NO.: TS8050000
EFFECTIVE: 05/05/86	REVISION #01
PRECAUTIONARY LABELLING BAKER SAF-T-DATA(TM) SYSTEM HEALTH - 0 NONE FLAMMABILITY - 0 NONE REACTIVITY - 0 NONE CONTACT - 1 SLIGHT HAZARD RATINGS ARE 0 TO 4 (0 = NO HAZARD; 4 LABORATORY PROTECTIVE EQUIPMENT: SAFETY PRECAUTIONARY LABEL STATEMENTS CAUTION MAY CAUSE IRRITATION DURING USE AVOID CONTACT WITH EYES, SKIN, HANDLING. WHEN NOT IN USE KEEP IN TIGHTLE SAF-T-DATA(TM) STORAGE COLOR CODE: ORAN	CLOTHING. WASH THOROUGHLY AFTE
2 - HAZARDOUS COMPONENTS	
COMPONENT % CAS NO. NOT APPLICABLE	
7 DILVOYO A 7 TO 1	
3 - PHYSICAL DATA	*
3 - PHYSICAL DATA FOILING POINT: 1437 C (2619 F) VAPOR PRESE MELTING POINT: 711 C (1312 F) VAPOR DENSE PECIFIC GRAVITY: 1.98 EVAPORATION R (H2O=1) (BUTYL ACETATE=1) OLUBILITY(H2O): APPRECIABLE (MORE THAN 187) PPEARANCE & ODOR: WHITE CRYSTALS OR POW	SURE(MM HG): N/A SITY(AIR=1): N/A ATE: N/A
OILING POINT: 1437 C (2619 F) VAPOR PRESSELTING POINT: 711 C (1312 F) VAPOR DENSE PECIFIC GRAVITY: 1.98 EVAPORATION R (H2O=1) (BUTYL ACETATE=1) OLUBILITY(H2O): APPRECIABLE (MORE THAN 1-PPEARANCE & ODOR: WHITE CRYSTALS OR POWAGE OF THE PRESSEL OF POWAGE OF THE PRESSEL OF POWAGE OF THE PRESSEL OF THE PRES	SURE(MM HG): N/A SITY(AIR=1): N/A ATE: N/A 0%) % VOLATILES BY VOLUME: 0 VDER.
OILING POINT: 1437 C (2619 F) VAPOR PRESSELTING POINT: 711 C (1312 F) VAPOR DENSE PECIFIC GRAVITY: 1.98 EVAPORATION R (H2O=1) (BUTYL ACETATE=1) OLUBILITY(H2O): APPRECIABLE (MORE THAN 1-PPEARANCE & ODOR: WHITE CRYSTALS OR POWAGE OF THE PRESSEL OF POWAGE OF THE PRESSEL OF POWAGE OF THE PRESSEL OF THE PRES	SURE(MM HG): N/A SITY(AIR=1): N/A ATE: N/A 0 %) % VOLATILES BY VOLUME: 0 VDER. /A % SURROUNDING FIRE. VE EQUIPMENT AND SELF-CONTAINED OPERATED IN POSITIVE PRESSURE MODE.

IOXICITY: LD50 (ORAL-RAT)(MG/KG) - 3020 LD50 (ORAL-MOUSE)(MG/KG) - 383 LD50 (IPR-RAT)(MG/KG) - 660 LD50 (IV-RAT) (MG/KG) - 39 CARCINOGENICITY: NTP: NO IARC: NO Z LIST: NO OSHA REG: NO EFFECTS OF OVEREXPOSURE DUST MAY IRRITATE SKIN OR EYES. INHALATION OF DUST MAY CAUSE IRRITATION TO UPPER RESPIRATORY TRACT. CONTACT MAY CAUSE MODERATE EYE IRRITATION. TARGET ORGANS: NONE IDENTIFIED MEDICAL CONDITIONS GENERALLY AGGRAVATED BY EXPOSURE: NONE IDENTIFIED ROUTES OF ENTRY: NONE INDICATED EMERGENCY AND FIRST AID PROCEDURES INGESTION: IF SWALLOWED AND THE PERSON IS CONSCIOUS, IMMEDIATELY GIVE LARGE AMOUNTS OF WATER. GET MEDICAL ATTENTION. INHALATION: IF A PERSON BREATHES IN LARGE AMOUNTS, MOVE THE EXPOSED PERSON TO FRESH AIR. GET MEDICAL ATTENTION. EYE CONTACT: IMMEDIATELY FLUSH WITH PLENTY OF WATER FOR AT LEAST 15 MINUTES. GET MEDICAL ATTENTION. SKIN CONTACT: IMMEDIATELY WASH WITH PLENTY OF SOAP AND WATER FOR AT LEAST
15 MINUTES.
6 - REACTIVITY DATA
STABILITY: STABLE HAZARDOUS POLYMERIZATION: WILL NOT OCCUR CONDITIONS TO AVOID: NONE DOCUMENTED
7 - SPILL AND DISPOSAL PROCEDURES
STEPS TO BE TAKEN IN THE EVENT OF A SPILL OR DISCHARGE WEAR SUITABLE PROTECTIVE CLOTHING. CAREFULLY SWEEP UP AND REMOVE. DISPOSAL PROCEDURE DISPOSE IN ACCORDANCE WITH ALL APPLICABLE FEDERAL, STATE, AND LOCAL ENVIRONMENTAL REGULATIONS.
8 - PROTECTIVE EQUIPMENT
VENTILATION: USE ADEQUATE GENERAL OR LOCAL EXHAUST VENTILATION TO KEEP FUME OR DUST LEVELS AS LOW AS POSSIBLE. RESPIRATORY PROTECTION: NONE REQUIRED WHERE ADEQUATE VENTILATION CONDITIONS EXIST. IF AIRBORNE CONCENTRATION IS HIGH, USE AN APPROPRIATE RESPIRATOR OR DUST MASK. EYE/SKIN PROTECTION: SAFETY GLASSES WITH SIDESHIELDS, GLOVES ARE RECOMMENDED.
9 - STORAGE AND HANDLING PRECAUTIONS
SAF-T-DATA(TM) STORAGE COLOR CODE: ORANGE (GENERAL STORAGE) SPECIAL PRECAUTIONS KEEP CONTAINER TIGHTLY CLOSED. SUITABLE FOR ANY GENERAL CHEMICAL STORAGE AREA
10 - TRANSPORTATION DATA AND ADDITIONAL INFORMATION
DOMESTIC (D.O.T.) PROPER SHIPPING NAME INTERNATIONAL (I.M.O.) PROPER SHIPPING NAME CHEMICALS, N.O.S. (NON-REGULATED)

AUG 94 CNUM=2578

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U.S. FOOD AND DRUG ADMINISTRATION FOOD ADDITIVE SAFETY PROFILE

TASSIUM CHLORIDE

MG/KG BW/DAY/PERSON LBS/YR	MG/KG BW/DAY/PERSON	
3.2485 3833333.333 87		LOGP:
HUMAN CONSUMPTION: MARKET DISAPPEARANCE: MARKET SURVEY:	JECFA: JECFA ADI: JECFA ESTABLISHED: LAST UPDATE:	DENSITY:
007447407 2578 NEW	1610	74.55
S#: SP#: PE:	AA#: AS#:	

A7 RUCTURE CATEGORIES:

MPONENTS:

NONYMS:

CHLORURE DE POTASSIUM POTASSIUM CHLORIDE (KCL)

ტ EMICAL FUNCTION: CHNICAL EFFECT:

NUTRIENT SUPPLEMENT FLAVOR ENHANCER FLAVORING AGENT OR ADJUVANT MALTING OR FERMENTING AID STABILIZER OR THICKENER ENZYME PROCESSING AID PH CONTROL AGENT

182.5622 184.1622 166.110 150.141

REG NUMBERS:

VIMUM TESTING LEVEL: 3

AMENTS: STUDIES 1-6 FROM SCOGS-102

ACUTE TOXICITY INFORMATION < 7:

1 RAT JDY: 3CIES:

SOURCE: ARCH INT PHARMACODYN THER 133:275-283
YEAR: 1961
LD50: 3020
MG/KG BW
MENTS: STUDY 1 LD50 = 3020 MG/KG FOR FEMALES

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CNUM=2578

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STUDY 8 LD50 = 3100 MG/KG
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MOUSE UDY: ECIES:

SOURCE: GRM 000091 25:5089 YEAR: 1974 LD50: 2350 MG/KG BW

MMENTS:

THAN ACUTE) ORAL TOXICITY STUDIES (OTHER .. 6 ≿

COMPLETENESS: SHORT TERM DOG JDY:
PE:
ECIES:
RATION:
FECTS:
FES:

SOURCE: J LAB CLIN MED 75:729-741 YEAR: 1970 MG/KG BW/DAY LEL: > MG/KG BW/DAY HNEL: 195 MG/KG BW/DAY

21 DAYS NO EFFECTS

TEST COMPOUND ADMINISTERED IN ENTERIC-COATED TABLETS ONE DOSE LEVEL ONLY

SOURCE: J DAIRY SCI 63:82-85 YEAR: 1980 LEL: > MG/KG BW/DAY HNEL: 1811 MG/KG BW/DAY COMPLETENESS: C SHORT TERM

CATTLE 21 DAYS NO EFFECTS

JDY:
PE:
3CIES:
RATION:
FES:
FES:

MALES ONLY DECREASED FOOD INTAKE AT 1811 MG/KG-MAY BE DUE TO A PALATABILITY PROBLEM

SOURCE: IZV AKAD NAUK ARM SSR BIOL SKH NAUKI 8:59-62 YEAR: 1955 LEL: 2000 MG/KG BW/DAY COMPLETENESS: SHORT TERM SHEEP

YEAR: LEL: HNEL:

1 DAYS MOTOR ACTIVITY DECREASE URINE VOLUME INCREASE SCIES: VATION:

res: aments:

2/2 SHEEP DIED AT 4000 MG/KG INCREASED RESPIRATION AND PULSE RATES AT 2000 AND 3000 MG/KG EFFECTS REVERSIBLE AFTER 6 HOURS

SOURCE: AM J PHYSIOL 166:273-276 YEAR: 1951 LEL: 5250 MG/KG BW/DAY HNEL: COMPLETENESS: C SC SUBCHRONIC RODENT YE YE S: RAT NO. 105 DAYS ORGAN WEIGHT DECREASE WATER INTAKE INCREASE CELLULAR HYPERTROPHY

JDY:
PE:
SCIES:
RATION:
PECTS:

ES:

ADRENAL GLAND